days at 30° . The intermediate produced yielded 2.3 g. of purified sapogenin. This was an over-all yield of 23% of the sapogenin which should have been obtained from this sample by using the regular procedure.

The remainder of the extract, containing saponins extracted from 1 kg. leaves, was adjusted with concd. HCl to pH 1.5 and refluxed 4 hours. From the resulting partial hydrolysis solids 6.2 g. of sapogenin was recovered, or 107% of the yield by the regular procedure. The data are given in Table II.

Acknowledgments.—The authors wish to express their sincere appreciation to James Garvin, Robert Pierce and Walter Rumph for preparing plant extracts; to Harriet C. Amsterdam, Theodore Perlstein and George Eppley for technical assistance in recovering sapogenins; to Charles Fenske, Howard Jones and C. Roland Eddy for infrared data; to Kathryn M. Zbinden and Clyde L. Ogg for microanalyses; to Jonathan W. White, Jr., and William L. Porter for assistance with sugar chromatography; and to A. M. Herbsman, N. B. Stepp, T. R. Stepp and Herschel Hunt for supplying the plant material.

PHILADELPHIA, PA.

[CONTRIBUTION FROM THE DEPARTMENT OF BIOCHEMISTRY, PURDUE UNIVERSITY]

Enzymatic Hydrolysis of Xylan¹

BY ROY L. WHISTLER AND EDWARD MASAK, JR.

RECEIVED SEPTEMBER 9, 1954

An enzyme preparation from the medium of *Aspergillus foetidus* hydrolyzed corn cob xylan primarily to xylobiose with no production of p-xylose. An extract of the mold mycelium was separated on a cation-exchange resin into two fractions; one which hydrolyzed xylan and xyloöligosaccharides to p-xylose without producing significant amounts of transient oligo-saccharides and one which hydrolyzed xylan to a mixture of p-xylose and xyloöligosaccharides. Fractionation of the mycelium extract on either celite or carboxymethylcellulose produced a fraction which did not hydrolyze xylan but did hydrolyze hemicellulose-B.

Xylan is hydrolyzed by animals,² plants,³ and by a large number of microörganisms.⁴ Hydrolysis of xylan by microörganisms was probably first examined by Hoppe-Seyler.⁵ Sorenson⁶ recently gave evidence that certain bacteria act on xylan to produce xyloöligosaccharides and D-xylose.

Since the use of specific xylan splitting enzymes would be of great assistance in examining the structure of the hemicelluloses, this Laboratory has undertaken the separation of the enzymes of *Aspergillus foetidus* into preparations with specific hydrolytic actions. Herein is described the initial separations achieved.

An extract of *A. foetidus* mycelium hydrolyzes xylan to D-xylose and xyloöligosaccharides as indicated by chromatographic comparison of the products with authentic samples.⁷ This enzyme preparation can be separated on a cation exchange resin into two fractions. A fraction (I) splits xylan to D-xylose, cellulose to D-glucose and hydrolyzes hemicellulose-B. Fraction II splits xylan to D-xylose and xyloöligosaccharides, cellulose to D-glucose and cellobiose but does not hydrolyze hemicellulose-B. A further fractionation of fraction I on a column of carboxymethylcellulose yields a portion (IA) that like fraction I hydrolyzes xylan, cellulose

(1) Journal Paper No. 804 of the Purdue Agricultural Experiment Station.

(2) M. Ehrenstein, Helv. Chim. Acta, 9, 332 (1926); C.A., 20, 2484
 (1926); H. Iwata, J. Agr. Chem. Soc. Japan, 11, 1017 (1935); C. A., 30, 2622 (1936).

(3) W. Voss and G. Butter, Ann., 534, 161 (1938); C. A., 32, 8772 (1938).

(4) F. B. Smith, Rept. Agr. Research, 1934, 129 (1934); C. A., 30, 1922 (1936); H. Iwata, J. Agr. Chem. Soc. Japan, 13, 978 (1937);
C. A., 32, 1739 (1938); W. H. Fuller and A. G. Norman, J. Back., 46, 281 (1943); K. Yamafuji and M. Inaoka, J. Agr. Chem. Soc. Japan, 23, 502 (1950); C. A., 45, 6709 (1951).

(5) F. Hoppe-Seyler, Z. physiol. Chem., 13, 66 (1889).

(6) H. Sorenson, Nature, 172, 305 (1953).

(7) R. L. Whistler and C. C. Tu, THIS JOURNAL, 74, 3609 (1952).

and hemicellulose-B and a portion (IB) that does not hydrolyze xylan and cellulose but does hydrolyze hemicellulose-B.

The mold mycelium extract can also be separated on columns of either celite or carboxymethylcellulose into a fraction of unchanged activity which hydrolyzes xylan, cellulose and hemicellulose-B, and another fraction which hydrolyzes hemicellulose-B but does not hydrolyze xylan or cellulose.

Fractional precipitation of the medium with ammonium sulfate yields a preparation which hydrolyzes xylan to xylobiose and higher xyloöligosaccharides but produces no noticeable p-xylose. This "xylo-dextrinase" has no effect on cellulose.

It is recognized that chromatographic comparisons do not constitute absolute identifications of Dxylose and xyloöligosaccharides. However such rapid and simple chromatographic comparisons with authentic specimens provide strong indications of the components present. This is particularly true in examining the enzymatic hydrolysis products of xylan because the structure of the polysaccharide is fairly well established and because its acid hydrolytic products have been well characterized as crystalline substances.⁷

Experimental

Materials.—Xylan was prepared from corn cob as previously described.⁸ Uronic acid: 2.9%, $[\alpha]^{25}D - 116.3^{\circ}$ (*c* 1.0 in 1 *N* NaOH); ash, 0.45%; molecular weight by light scattering, 34,500; D.P., 261. Hemicellulose-B was also prepared as previously described. Swollen or "reactive" cellulose was prepared according to the phosphoric acid procedures outlined by Walseth.^{9,10} The mold, *A. foetidus*¹¹ was

(8) R. L. Whistler, J. Bachrach and D. R. Bowman, Arch. Biochem., 19, 25 (1948).

(9) C. S. Walseth, Ph.D. Dissertation, Lawrence College, 1948.
(10) R. L. Whistler and C. L. Smart, THIS JOURNAL, 75, 1916 (1953).

(11) C. Thom and K. B. Raper, "A Manual of the Aspergilli," The Williams and Wilkins Company, Baltimore, Maryland, 1945, p. 219.

grown on a liquid medium of 1.5% D-glucose, 3.0% bacto-

peptone and inorganic salts.¹² Identification of Oligosaccharides.—Paper chromato-graphic analyses of the hydrolyzates of xylan by the various enzyme preparations were compared to authentic xylo-oligosaccharides previously isolated and characterized.⁷ Descending chromatograms were developed at 25° on Whatman No. 1 paper with a basic solvent of ethyl acetate, pyridine and water in a ratio of 10:4:3 and with an acidic solvent of ethyl acetate, acetic acid, formic acid and water in a ratio of 18:3:1:4. The papers were dried and sprayed with *p*-anisidine hydrochloride.¹³ Colored areas resulting from interaction of spray reagent and sugars appeared after heat-ing in an oven for 1 minute at 120°. The oligosaccharides in the enzyme hydrolyzate correspond in R_x values (ratio of distance traveled to the distance D-xylose travels in the same time) with authentic crystalline xyloöligosaccharides.⁷ $R_{\mathbf{x}}$ values of known xyloöligosaccharides in the above-mentioned acidic and basic developing agents are given in Table

TABLE I

R_{x} Values of Xyloöligosaccharides

| | Eluant | | |
|--------------|--------------|-------------|--|
| Sugars | Acidic | Basic | |
| Xylobiose | 0.380-0.400 | 0.678-0.690 | |
| Xylotriose | .145 - 0.152 | .427-0.440 | |
| Xylotetraose | .038-0.050 | .236-0.242 | |
| Xylopentaose | .0013 | .110 | |
| Xylohexaose | | .0395 | |
| Xyloheptaose | | .0285 | |
| Xyloöctaose | | .0157 | |

Enzyme Fractionation .- Two liters of autoclaved medium was inoculated with mold spores. After four days at 25° before the mold had begun to form spores, the medium was drained from the mycelium, suction filtered through Whatman No. 57 paper and 80% saturated with ammonium sulfate. After 12 hours, the precipitate was removed by cen-trifugation and dialyzed at 4° in a collodion sac against frequently changed distilled water for 4 days. The resulting enzyme solution was freeze-dried to yield approximately 0.92 g. of a tan colored powder. Its nitrogen content of 9.6%

indicated 60.1% protein present. Thirty mg. of this enzyme preparation was added to 75 Thirty mg. of this enzyme preparation was added to 75 mg. of xylan in 15 ml. of pH 4.0 buffer solution at 37°. During the hydrolysis, aliquots were regularly taken for chromatographic analysis. In none of these was there any evidence of D-xylose; xylobiose appeared to be the smallest carbohydrate molecule present. The hydrolysis was slow, taking 24 hours to degrade 9.25% of the substrate. The reducing values, calculated as D-xylose, were used to determine the extent of hydrolysis.¹⁴ The enzyme had no noticeable action on cellulose. However, an enzyme preparation obtained in a similar manner from the medium of a spored 7-day old mold hydrolyzed xylan to D-xylose and spored 7-day old mold hydrolyzed xylan to D-xylose and xyloöligosaccharides. This change in activity may be accounted for by autolysis of cells which might release a greater amount of enzymes into the medium at this stage.

The mycelium of a 4-day old A. foetidus culture was ground in a blendor with fine clean sand and a sufficient amount of 5% aqueous sodium chloride solution to produce amount of 3% aqueous sodium chloride solution to produce a thick slurry. The cell-free extract was filtered through Whatman No. 1 paper on a buchner funnel and the residue washed with fresh 5% sodium chloride solution. The fil-trate was 80% saturated with ammonium sulfate and allowed to stand at 4° for 12 hours. The precipitate was centri-fuged and dialyzed at 4° in a colloidon sac against distilled water for 4 days. Freeze-drying produced a tan powder which hydrolyzed xylan to D-xylose and xyloöligosacchar-ides. The nitrogen content of the powder was 9.1% indi ides. The nitrogen content of the powder was 9.1% indicating a protein content of 56.8%. A column (2 × 13 cm.) of 140 mesh IRC-50 resin in the

hydrogen form was washed with MacIlvaine's citric acid-

phosphate buffer at pH 5.8.15 After addition of 0.6 g. of the dried cell-free mycelium extract, the column was eluted with pH 6.5 buffer at 20° at the rate of 1 ml. per 3 minutes. One-ml. fractions were collected and added to 1.0 g. of dry ashless xylan or hemicellulose-B in 11 ml. of pH 4.5 buffer. The hydrolysis proceeded at 37° for 48 hours. After this period the reducing values (as D-xylose) of the substrates were determined and chromatograms made. Chromatograms of the xylan hydrolyzates produced by the first 12 to 22 ml. of eluate, fraction I, showed the production of Dxylose only, while all other active eluates II produced Dxylose and other xyloöligosaccharides. A plot of the re-ducing values for xylan against the ml. of eluant gave a bimodal curve, with the major apex corresponding to the hydrolytic action of fraction I and the minor apex corresponding to the hydrolytic action of fraction II. Chromatographic evidence and reducing values indicated that only fraction I hydrolyzed hemicellulose-B. Fraction I also hydrolyzed cellulose to D-glucose while fraction II hydrolyzed cellulose to D-glucose and cellobiose.

The separation was repeated on additional cell-free extract. The fractions with p-xylose producing activity were combined and freeze-dried to give 0.30 g. of a tan powder (I). This enzyme preparation contained 9.8% nitrogen indicating 61.3% protein. Fraction I was separated on a 1×13 cm. column of car-

boxymethylcellulose¹⁸ (degree of substitution, 0.563) previously washed with pH 5.1 mono and dibasic sodium phosphate buffer. Three hundred mg. of fraction I was dissolved in 3 ml. of water, placed on the column and eluted with pH 6.5 phosphate buffer. The elution was done as rapidly as possible as the enzyme preparation can hydrolyze car-boxymethylcellulose.¹⁷ Two fractions were separated. One fraction (IA) found in the first 11 ml. of eluate retained the original activity of I and hydrolyzed xylan to D-xylose and hemicellulose-B to p-xylose and various oligosaccharides. A second fraction (IB) eluted after the first 11 ml. hydrolyzed hemicellulose-B to oligosaccharides with no

evidence of D-xylose. Xylan is not affected by this fraction. Optimum pH of Dried Enzyme Preparations.—Thirty mg. individual samples of the mycelium extract and of the frac-tion I and 60-mg, samples of the medium extract were each added to 50-mg, individual samples of xylan in 5 ml. of water. Five-ml. aliquots of six different MacIlvaine buffers were added to each sample at 37°. After 5 hours reducing values were determined. By assuming completion of the hydrolysis when all the xylan was hydrolyzed to D-xylose, the ratio of the reducing value to the final reducing value was taken as a measure of the extent of hydrolysis.

| TABLE | II |
|-------|----|
|-------|----|

EFFECT OF pH ON ENZYMATIC ACTIVITY

| Source of enzyme preparation | pH | Hydrolysis, $\%$ |
|---------------------------------|-----|------------------|
| Medium | 2.8 | 14.8 |
| | 3.0 | 15.8 |
| | 3.4 | 17.2 |
| | 3.6 | 16.6 |
| | 4.0 | 15.2 |
| | 5.0 | 12.2 |
| Mycelium | 3.0 | 13.2 |
| | 4.0 | 16.4 |
| | 4.5 | 18.4 |
| | 5.0 | 16.0 |
| | 5.5 | 13.0 |
| Fraction I | 3.0 | 17.8 |
| | 3.6 | 19.7 |
| | 3.8 | 20.4 |
| | 4.0 | 19.8 |
| | 4.5 | 18.4 |
| | 5.0 | 17.2 |
| | | |

(15) N. K. Boardman and S. M. Partridge, J. Polymer Sci., 12, 281 (1954).

(16) H. A. Sober and E. A. Peterson, THIS JOURNAL, 76, 1711 (1954).

(17) S. O. Pehrson, Svensk Papperstidn., 53, 726 (1950); C. A., 45, 7789 (1951).

⁽¹²⁾ H. F. Smyth and W. L. Obald, "Industrial Microbiology," The Williams and Wilkins Company, Baltimore, Maryland, 1930, p. 46. (13) L. Hough, J. K. N. Jones and W. H. Wadman, J. Chem. Soc.,

^{1702 (1950).}

⁽¹⁴⁾ P. A. Shaffer and A. F. Hartman, J. Biol. Chem., 45, 349 (1921).

The medium extract gave an optimum hydrolysis at pH 3.4, the mycelium extract at pH 4.5 and the fraction I at pH 3.8 (Table II).

Optimum Temperature.—Samples of enzyme preparations were weighed out similar to those used previously to determine the optimum pH. Ten ml. of pH 3.4, 3.8 and 4.5 buffers were added to the medium extract, the fraction I and the mycelium extract, respectively. These enzyme solutions were individually added to 50 mg. of xylan. Mixtures of each enzyme preparation were held at 25, 37, 45 or 55° for 5 hours. At this period, reducing values determined the optimum temperature for the mycelium and medium extracts as 37° and the fraction I as 45° (Table III).

TABLE III

Effect of Temperature on Enzymatic Activity

| Source of enzyme preparation | Temp., °C. | Hydrolysis, $\%$ |
|---------------------------------|------------|------------------|
| Medium | 25 | 8.0 |
| | 37 | 17.2 |
| | 45 | 15.2 |
| | 55 | Nil |
| Mycelium | 25 | 12.0 |
| | 37 | 18.8 |
| | 45 | 16.4 |
| | 55 | 9.7 |
| Fraction I | 25 | 22.0 |
| | 37 | 30.8 |
| | 45 | 32.2 |
| | 55 | 28.2 |
| | | |

Heat Inactivation .--- Solutions containing 100 mg. of dried cell-free mycelium extract preparation or 100 mg. of dried medium extract preparation in 5 ml. of water were heated at 65°. At intervals 0.5 ml. of the solution was added to 50-mg. samples of xylan in 2 ml. of pH 4.0 buffer at 37° After standing for 20 hours chromatograms were made to determine whether hydrolysis had occurred. Inactivation of the medium extract occurred after 5 minutes heating. The mycelium extract that had been heated for less than 30 minutes hydrolyzed xylan to D-xylose and the xyloöligosaccharides as indicated by paper chromatography. On the other hand the mycelium extract that had been heated for longer than 30 minutes hydrolyzed xylan to D-xylose only. Fraction I is not inactivated after 6 hours heating and thus by heating the mycelium extract at 65° for 30 minutes the fraction I activity can be separated from the fraction II activity.

Enzyme Action on Oligosaccharides.—Individual samples of 25 mg. of pure xylobiose, xylotriose and xylotetraose were each dissolved in 2 ml. of water. To one series 2 ml. of pH 3.4 buffer and 25 mg. of dried medium extract were added. To another series 10 mg. of dried fraction I and 2 ml. of pH 3.8 buffer were added. Both series were maintained at 37°. The course of hydrolysis was followed by taking hourly aliquots of 10 λ (microliters) for chromatographic analysis.

Search for Phosphorylated Sugars.—Phosphorylated sugars may be detected on a paper chromatogram by a variety of color tests. One of the most satisfactory procedures is Gutzeit's method as modified by Whistler and Smart.¹⁰ This method did not indicate the presence of phosphoryl-

TABLE IV

ENZYME ACTION ON XYLOÖLIGOSACCHARIDES

| Oligosaccharide substrate | End product(s) | time elapsed, hr. |
|------------------------------|---|--|
| Xylobiose | No reaction | |
| Xylotriose | No reaction | •• |
| Xylotetraose | Xylobio s e | 18 |
| Xylobiose | D-Xylose | 8 |
| Xylotriose | Xylobiose and D- | |
| | xylose | 4 |
| | D-Xylose | 8 |
| Xylotetraose | Xylotriose, xylobi- ose and D-xylose Xylobiose and D- | 4 |
| | xylose | 7 |
| | D-Xylose | 8 |
| | substrate Xylobiose Xylotriose Xylotetraose Xylobiose Xylotriose | substrateEnd product(s)XylobioseNo reactionXylotrioseNo reactionXylotetraoseXylobioseXylobioseD-XyloseXylotrioseXylobiose and D-xylotetraoseD-XyloseXylotetraoseXylotriose, xylobioseXylotetraoseXylotriose, xylobioseXylotetraoseXylotriose, xylobioseXylotetraoseXylotriose, xylobioseXylotetraoseXylotriose, xylobioseXylobiose and D-xyloseXylobiose and D-xylobioseXylobiose and D-XylobioseXylobioseXylobioseXylobioseXylobioseXylobioseXylobioseXylobiose |

ated sugars either as intermediates or end products in the hydrolysis of xylan or in the hydrolysis of the xylooligosaccharides by any of the enzymes described.

Separation of Xylan-splitting from Hemicellulose-Bsplitting Enzymes.—The enzymes responsible for the hydrolysis of hemicellulose-B were separated from the enzymes responsible for xylan hydrolysis by two methods.

The first method was a modification of Van Sumere's.¹⁸ Three grams of powdered mycelium extract was dissolved in 75 ml. of water 66% saturated with ammonium sulfate. This solution was mixed with 15 g. of celite, placed in a glass tube, 2 cm. in diameter, and allowed to settle at 20°. The column was eluted with 33% saturated ammonium sulfate solution at a rate of 5 ml./15 minutes. Each 5 ml. of effluent was added to 13 ml. of β H 4.5 solutions of xylan or cellulose or hemicellulose-B at 37° (60° for the cellulose mixture). After 24 hours the reducing values (calculated as D-xylose for xylan and hemicellulose-B and D-glucose for cellulose) were determined. The fraction III eluted in the first 115 ml. was active against xylan, cellulose and hemicellulose-B. The fraction IV eluted after this point was active against hemicellulose-B only. A second method of communications of the cellulose and the first 115

A second method of enzyme separation involved the use of carboxymethylcellulose prepared as described above. Using a 300-mg. sample of mycelium extract on a pH 5.1 buffered column and eluting with pH 6.5 buffer, produced a fraction III in the first 9 ml. which was active against xylan, cellulose and hemicellulose-B. Upon further elution a second fraction IV was separated which hydrolyzed only hemicellulose-B.

Acknowledgment.—The authors are pleased to acknowledge their indebtedness to Professors C. L. Porter and H. W. Reuszer of Purdue University for identifying the mold, Mr. J. W. Groce of this department for assisting with the hemicellulose-B portion of the experimental work and to Dr. F. R. Senti of the Northern Utilization Branch, Peoria, Illinois, for the determination of the molecular weight of the xylan.

LAFAVETTE, INDIANA

(18) C. Van Sumere, Naturwissenshaften, 22, 583 (1953).